POZNAN UNIVERSITY OF TECHNOLOGY



EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS) pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

COURSE DESCRIPTION CARD - SYLLABUS

Course name		
Fundamentals of Electrochemical Te	chnology	
		Course
Field of study		Year/Semester
Environmental Protection Technologies Area of study (specialization)		III/5
		Profile of study
-		general academic
Level of study		Course offered in
First-cycle studies		polish
Form of study		Requirements
full-time		compulsory
		Number of hours
Lecture	Laboratory classes	Other (e.g. online)
30	30	0
Tutorials	Projects/seminars	
0	0	
Number of credit points		
5		
		Lecturers
Responsible for the course/lecturer:		Responsible for the course/lecturer:

dr hab. Piotr Krawczyk, prof. PP

Prerequisites

Student has a ordered knowledge of mathematics and physical chemistry and he also has ability to use the basic techniques in a laboratory scale. Student understand the need for further education and enhance of professional and personal competences.

Course objective

The aim of the course is to familiarize students with an overview of technical electrochemistry methods and develop skills for their practical application.

Course-related learning outcomes

Knowledge

1. The knowledge in the field of basics of electrochemical processes –[K_W12],

2. The knowledge in the field of various electrochemical technologies and apparature used-[K_W10],

3. The knowledge in the field of related fields –[K_W06].

Skills

1. The student has the ability to selection of measurement techniques –[K_U12],



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2. The student has the ability to perform the charcterization of materials used in electrochemistry – [K_U13],

2. The student has the ability to acquired the information necessary to conduct the investigations – [K_U01].

Social competences

1. The student understands the need for self-study and improvement of their professional competence –[K_K01],

2. Student can act and cooperate in the group accepting different roles –[K_K03].

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Rating of written answers within the subjects related to the theme of the practical classes.

Checking of current knowledge and practical skills, the ability to conduct experiments corectly during laboratory classes. Performing all laboratory exercises provided for the study program. Final mark of the laboratory class will correspond to the mean of the sum of the above.

In the case of on-line classes, the knowledge check will be carried out in the form of a test consisting of 3 - 5 questions for each exercise and report for the given experimental data.

The knowledge acquired during the lecture is verified by a written final exam in the subject consisting of 3 questions. Passing threshold will correspond to 51% of the maximum number of points.

In the case of on-line classes, the exam will take the form of a test consisting of 20 test questions and five open questions. Passing threshold: 51% of the maximum number of points.

Programme content

- 1. The principles of electrochemical processes.
- 2. Electrodes balances.
- 3. The kinetics of electrode processes.
- 4. The selected electrochemical processes.
- 5. The processes based on the electrochemical processes,
- 6. Examples of regeneration processes apply in electrochemistry.
- 7. Electrochemicl processes associated with the generation, conversion and storage of electrical energy.

Teaching methods

Lecture, problem lecture, explanation, didactic discussion, classes, project method, laboratory exercises

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Basic

- 1. A. Kisza Elektrochemia cz. I i II (Jonika i Elektrodyka) WNT, W-wa, 2001,
- 2. R. Dylewski, W. Gniot, M. Gonet, Elektrochemia przemysłowa, Wyd. Politechniki Śląskiej, 1999,
- 3. A. Czerwiński, "Ogniwa, akumulatory, baterie", WNT, W-wa, 1999,
- 4. C. G. Zoski praca zb., Handbook of Electrochemistry, Elsevier, 2007,
- 5. A. Ciszewski, Technologia chemiczna. Procesy elektrochemiczne, Wyd. Politechniki Poznańskiej, 2008.

Additional

1. A.V. da Rosa, Fundamentals of Renewable Energy Processes, Elsevier/Academic Press, 1990,

2. H. Scholl, T. Błaszczyk, P. Krzyczmonik, Elektrochemia, Wyd. Uniwersytetu Łódzkiego, 1998.

Breakdown of average student's workload

	Hours	ECTS
Total workload	140	5,0
Classes requiring direct contact with the teacher	70	2,5
Student's own work (literature studies, preparation for	70	2,5
laboratory classes/tutorials, preparation for tests/exam, project		
preparation) ¹		

¹ delete or add other activities as appropriate